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Chemists, Engineers, And Physicists By V. Adrian
Parsegian**

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van der Waals force A weak force of attraction between electrically neutral molecules that collide with or pass very close to each other. The van der Waals force is

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van der Waals force. n. A weak attractive force between atoms or nonpolar molecules caused by a temporary change in dipole moment arising from a brief shift of

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Mar 16, 2008 Van der Waals forces include momentary attractions between molecules, diatomic free elements, and individual atoms. They differ from covalent and ionic

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Van der Waals forces are driven by induced electrical interactions between two or more atoms or molecules that are very close to each other. Van der Waals interaction

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V. Adrian Parsegian is the author of Van Der Waals Forces (3.00 avg rating, 2 ratings, 0 reviews, published 2005) register; V. Adrian Parsegian Author profile

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Oct 22, 2006 In chemistry, the term Van der Waals force refers to a particular class of intermolecular forces. The term originally referred to all such forces, and this

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van der Waals' forces or van der Waals forces Word Origin plural noun, Physical Chemistry 1. weak, nonspecific forces between molecules. Origin of van der Waals

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It has been pointed out earlier in the beginning of this unit that the attractive forces existing between polar as well non-polar molecules are collectively called

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Definition of VAN DER WAALS FORCES: the relatively weak attractive forces that act on neutral atoms and molecules and that arise because of the electric polarization

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Van der Waals Equation. Van der Waals equation is required for special cases, such as non-ideal (real) gases, which is used to calculate an actual value.

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